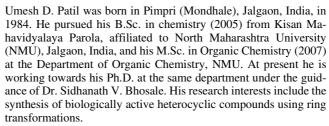
2880 SPOTLIGHT

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This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

Bestmann-Ohira Reagent: A Versatile Reagent in Organic Synthesis

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Dedicated to Professor S. L. Patil



Bestmann–Ohira reagent [(1-diazo-2-oxopropyl)phosphonate] can be prepared by the reaction of dimethyl-2-oxopropylphosphonate, TosN₃^{1a} or *p*-acetamidobenzene-sulfonyl azide, ^{1b} NaH, *t*-BuOK or Et₃N in benzene and THF. An alternative is the preparation using polymer-supported sulfonyl azide and *t*-BuOK in methylenchloride. ^{1c} The Bestmann–Ohiro reagent is widely used in the con-

version of primary alcohols, aldehydes, ketones, and amides into alkynes. Recently, it was employed in the synthesis of pyrazoles as well as 1,3-oxazoles.

Scheme 1

Abstracts

(A) The use of the Bestmann–Ohira reagent is an alternative to the Fritsch–Buttenberg–Wiechell-type rearrangement and the Corey–Fuchs procedure that allows the addition of the reagent to an aldehyde under mild reaction conditions, thus avoiding the use of a strong base under low-temperature conditions. The reaction works with alkyl and aryl as well as hindered aldehydes. In case of α,β -unsaturated aldehydes the main products were isolated as homopropargylic methyl ethers. The reaction can be easily performed under one-pot conditions. 2

(B) Hamilton D. Dickson and co-workers reported that esters and amides undergo reduction to the corresponding aldehydes using DIBAL-H followed by in situ conversion into terminal alkynes utilizing the Bestmann–Ohira reagent in good to excellent yields.³ Additionally, chiral nonracemic substrates undergo this transformation with complete preservation of stereochemical integrity.³

(C) E. Quesada et al. described the direct conversion of activated primary alcohols into terminal alkynes through a sequential one-pot, two-step process involving oxidation with manganese dioxide and then treatment with the Bestmann–Ohira reagent. This transformation proceeds efficiently (59–99% yield) under mild reaction conditions with a range of benzylic, heterocyclic, and propargylic alcohols.⁴

OH

1. MnO₂

2. Bestmann–Ohira reagent

$$X = NO_2$$

$$X = CO_2Me$$

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X = H, Br, NO₂

(D) 1,3-Dipolar cycloaddition of the anion of diethyl 1-diazomethylphosphonate, generated in situ from Bestmann-Ohira reagent, with conjugated nitroalkenes provided regioisomerically pure phosphonylpyrazoles in moderate to good yield. These pyrazoles are formed in one pot via spontaneous elimination of the nitro group.⁵

(E) D. Gong et. al. prepared a series of 4-phosphoryl-substituted 1,3oxazoles conveniently by reaction of Bestmann-Ohira reagent and aromatic nitriles in the presence of a catalytic amount of rhodium(II) acetate.6

(F) Oxidation of epoxy alcohol by Dess-Martin periodinane gave an aldehyde which was subsequently treated with the Ohiro reagent to give an epoxy alkyne in 93% yield.⁷

(G) Barrett and co-workers prepared ROMPgel-supported ethyl 1diazo-2-oxopropylphosphonate and employed this reagent in the conversion of a variety of aldehydes into terminal alkynes.8 The use of polymer-supported Ohiro reagent led to high product yield.

(H) The cycloaddition of diethyl 1-diazomethylphosphonate with benzylidine alkylamines afford (1-alkyl-5-phenyl-4,5-dihydro-1H-[1,2,3]triazol-4-yl)-phosphonic acid diethyl ester.

$$N_2$$
CHPO(OEt)₂ + Ph H $MeOH$ $r.t., 3 d$ Ph N N N

(I) Gilbert et al. described the base-promoted reaction of dimethyl (diazomethyl)phosphonate with aldehydes and aryl ketones at low temperature. The alkynes are obtained in modest to excellent yields.10

(J) The Bestmann-Ohira reagent itself was used for the preparation of enol ethers or alkynes from carbonyl compounds in the presence of excess potassium carbonate and methanol.¹¹

$$\begin{array}{c} & & & \\ & &$$

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