

Enantioselective β -Lactone Formation from Phenyl diazoacetates via Catalytic Intramolecular Carbon-Hydrogen Insertion

Michael P. Doyle,* Eric J. May

Department of Chemistry, University of Arizona, Tucson, Arizona 85723, USA

Fax +1 520 6264815; E-mail: mdoyle@u.arizona.edu

Received 26 January 2001

Abstract: Dirhodium(II) catalysts with chiral carboxylate or carboxamidate effectively promote β -lactone formation from phenyl diazoacetates in high yield and with up to 63% ee.

Key words: enantioselective carbon-hydrogen insertion, chiral dirhodium(II) catalysts, phenyl diazoacetates, β -lactones

Dirhodium(II) catalyzed intramolecular carbon-hydrogen insertion reactions originating with diazocarbonyl compounds have enjoyed wide popularity for the synthesis of cycloalkanones, lactones, and lactams.¹⁻⁴ They exhibit a high preference for the formation of five-membered rings and, in the absence of conformational restrictions,⁵ reactivity follows the order tertiary > secondary >> primary.⁶ There are few examples of insertion reactions favoring ring sizes other than five in these reactions,^{1,7-9} even when electronic influences would justify them.¹⁰ Recently, Davies and coworkers have demonstrated that aryldiazoacetates exhibit much higher levels of selectivity in C-H insertion reactions.¹¹ Based on this report and other indicators of reactivity/selectivity,¹ we have searched for carbon-hydrogen insertion reactions that could provide the formation of four-membered ring β -lactones in reasonable yields and with catalyst-directed enantiocontrol.

The first substrate tested was isopropyl phenyl diazoacetate, and we were surprised to find that the corresponding β -lactone was virtually the sole product in reactions that were catalyzed by rhodium acetate and by dirhodium(II)

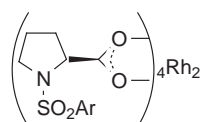
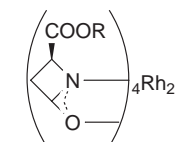
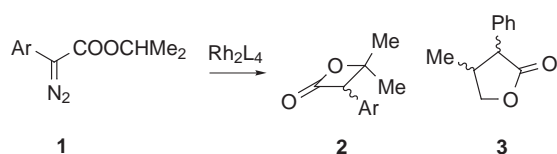
compounds **4** and **5** (Table 1). Here, insertion into the 3 °C-H bond was favored over insertion into one of six 1 °C-H bonds despite the additional strain introduced by formation of a four- rather than a five-membered ring. Traces of **3**, mainly the *trans*-disubstituted lactone, were observed, but the overall difference in reactivity could be estimated to be greater than 50:1. Products were identified by spectroscopic analysis with reference to literature reports of the same compounds.¹²

Table 1 Enantioselectivity in Carbon-Hydrogen Insertion Reactions of Isopropyl Phenyl diazoacetate^a

catalyst	yield, % ^b	ee, % ^c
	2 (Ar = Ph)	2 (Ar = Ph)
Rh ₂ (OAc) ₄	84	—
Rh ₂ (<i>S</i> -MEAZ) ₄ (4a)	83	33
Rh ₂ (<i>S</i> -IBAZ) ₄ (4b)	79	26
Rh ₂ (<i>S</i> -BNAZ) ₄ (4c)	66	30
Rh ₂ (<i>S</i> -CHAZ) ₄ (4d)	85	35
Rh ₂ (<i>S</i> -NEPAZ) ₄ (4e)	84	24
Rh ₂ (<i>S</i> -DOSP) ₄ (5)	86	36
Rh ₂ (<i>S</i> -DOSP) ₄ (5) ^d	78	41

^a Reactions were performed in refluxing CH₂Cl₂, unless specified otherwise, using 1.0 mol% of catalyst. ^b Yield of product after separation of catalyst (up to 70% yield after chromatographic purification).

^c Enantiomer separation and analyses were performed on a 25-cm, 4.6-mm (*R,R*)-WHELK-O column using 5% EtOAc in hexanes (8.2 and 9.0 min for the individual enantiomers). ^d Reaction performed in refluxing pentane.



4a: R = Me, Rh₂(4*S*-MEAZ)₄

4b: R = ^tBu, Rh₂(4*S*-IBAZ)₄

4c: R = Bn, Rh₂(4*S*-BNAZ)₄

4d: R = cC₆H₁₁, Rh₂(4*S*-CHAZ)₄

4e: R = CH₂CMe₃, Rh₂(4*S*-NEPAZ)₄

5: Ar = *p*-C₁₂H₂₅C₆H₄, Rh₂(*S*-DOSP)₄

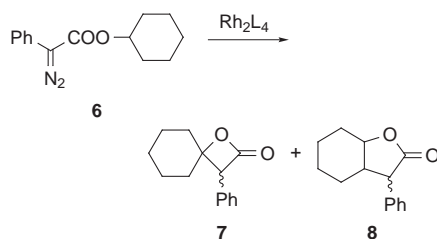
Scheme 1

Reactions catalyzed by chiral dirhodium(II) compounds, either the Rh₂(*S*-DOSP)₄ catalyst of Davies⁴ or our own chiral azetidinone-ligated catalysts,^{13,14} generally resulted in β -lactone product in high yield but with only modest enantioselectivities. The use of Rh₂(*S*-DOSP)₄ produced product with the highest% ee value, especially when the reaction was performed in pentane. Reactions with azetidinone-ligated catalysts in pentane provided no obvious advantages over reactions performed in CH₂Cl₂.

An attempt was made to determine the electronic influence of aryl substituents from aryldiazoacetates on enantiocontrol. However, significantly lower product yields were obtained with Ar = *p*-MeOC₆H₄ – a substituent that, based on published reports by Davies,^{11,15,16} we thought would lead to modest changes in enantioselectivity,% ee

values were considerably lower than those reported in Table 1 (33% ee with **5**). However, with Ar = *p*-MeC₆H₄, Rh₂(*S*-DOSP)₄ gave the corresponding β-lactone in 77% yield with 48% ee, but Rh₂(*S*-MEAZ)₄ gave product in lower yield (34%) and with lower enantioselectivity (27% ee). The reason for this apparent discrepancy is as yet unresolved.

Diazo decomposition of cyclohexyl diazoacetate produces the γ-lactone products virtually exclusively.⁷ The corresponding β-lactone, if formed at all, is a very minor product. In contrast, diazo decomposition of cyclohexyl phenyldiazoacetate **6** gives the corresponding β-lactone product **7**¹⁷ with near exclusivity (Table 2), and the catalyst had virtually no influence on regioselectivity. Here, enantioselectivities were higher than those obtained with isopropyl phenyldiazoacetate. Comparable results were obtained with *cis*-4-methylcyclohexyl phenyl-diazoacetate [with Rh₂(4*S*-MEAZ)₄ in CH₂Cl₂: 74% yield, 44% ee; with Rh₂(*S*-DOSP)₄ in pentane: 56% yield, 44% ee].



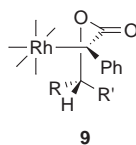
Scheme 2

Table 2 Enantioselectivity and Regioselectivity in Carbon-Hydrogen Insertion Reactions of Cyclohexyl Phenyldiazo-acetate^a

catalyst	yield, % ^b	ee, % ^c	
		7	7:8 ^d
Rh ₂ (OAc) ₄	55	—	98:2
Rh ₂ (<i>S</i> -MEAZ) ₄ (4a)	67	50	98:2
Rh ₂ (<i>S</i> -IBAZ) ₄ (4b)	66	51	97:3
Rh ₂ (<i>S</i> -NEPAZ) ₄ (4c)	65	42	97:3
Rh ₂ (<i>S</i> -DOSP) ₄ (5)	52	49	98:2
Rh ₂ (<i>S</i> -DOSP) ₄ (5) ^d	69	63	98:2

^a Reactions were performed as described in Table 1. ^b Product yield after separation of catalyst (up to 53% yield of **7** after chromatographic purification). ^c Enantiomer separation and analyses were performed on a 25-cm, 4.6-mm (R,R)-WHELK-O column using 5% EtOAc in hexanes (8.2 and 9.1 min for the individual enantiomers). ^d Determined by ¹H NMR of unique spectral regions for **7** (ref. 17) and **8** (ref. 18). ^e Reaction performed in refluxing pentane.

In each C–H insertion reaction presented thus far the chiral center that is generated is the original diazo-carbon atom with a probable transition state orientation that is depicted in **9**. When reactions of **1**, **6**, and their analogs R and R' are identical and, if insertion occurs as shown in **9**,



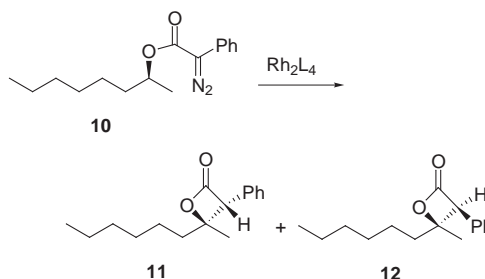
the configuration of the β-lactone product would be *R*. When R and R' are not identical, one can expect a diastereomeric product distribution that reflects the relative stabilities of attached carbene/catalyst configuration. This is effectually represented in results from diazo decomposition of (*S*)-(+)-2-octyl phenyldiazo-acetate (**10**) from which two diastereomeric β-lactone products (**11** and **12**)¹⁹ are formed in good yields (Table 3). γ-Lactone products were, at best, trace constituents of the reaction mixture. As expected, changing catalyst configurations, Rh₂(4*S*-MEAZ)₄ and Rh₂(4*R*-MEAZ)₄, resulted in modest, but measurable, differences in diastereoselection indicative of match/mismatch in catalyst-substrate interactions. Surprisingly, the **11:12** ratio with Rh₂(*S*-DOSP)₄ was opposite to that with the 4*S*-azetidinone-ligated dirhodium(II) catalysts, and the reason for this is unknown.

Table 3 Diastereoselectivity in Carbon-Hydrogen Insertion Reactions of (*S*)-(+)-2-Octyl Phenyldiazoacetate^a

catalyst	yield, %	ee, %
	11 + 12 ^b	11:12 ^c
Rh ₂ (OAc) ₄	91 ^d	37:63
Rh ₂ (<i>S</i> -MEAZ) ₄ (4a)	79	22:78
Rh ₂ (<i>R</i> -MEAZ) ₄ (<i>ent</i> - 4a)	60	53:47
Rh ₂ (<i>S</i> -IBAZ) ₄ (4b)	68	36:64
Rh ₂ (<i>S</i> -DOSP) ₄ (5) ^e	56	69:31

^a Reactions were performed as described in Table 1. ^b Yield after removal of catalyst. ^c Determined by ¹H NMR analysis and confirmed by GC on a SPB-5 column operated at 100 °C. ^d Column chromatography on silica (2.5–10% EtOAc in hexanes) allowed isolation of **11** (28% yield) and **12** (57% yield) as separate products. ^e Reaction performed in refluxing pentane.

Overall, dirhodium(II) catalysts are surprisingly selective



Scheme 3

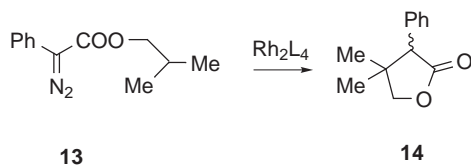
for β-lactone formation, affording these products in high

yield, although with modest enantiocontrol. However, when a tertiary C–H bond is available that could result in a γ -lactone product, as is the case with isobutyl phenyl diazoacetate (**13**), only the γ -lactone product is observed (Table 4). Here use of the chiral azetidinone-ligated catalysts gave comparable % ee values for the insertion product (**14**) to results from $\text{Rh}_2(\text{S-DOSP})_4$ in pentane. Clearly, the presence of a tertiary C–H bond directs C–H insertion with phenyl diazoacetates to a far greater extent than that found with diazoacetates alone.

Table 4 Enantioselectivity in Carbon-Hydrogen Insertion Reactions of Isobutyl Phenyl diazoacetate^a

catalyst	yield, %	ee, %
	14 ^b	14 ^c
$\text{Rh}_2(\text{OAc})_4$	79	—
$\text{Rh}_2(\text{S-MEAZ})_4$ (4a)	94	90
$\text{Rh}_2(\text{S-IBAZ})_4$ (4b)	89	84
$\text{Rh}_2(\text{S-DOSP})_4$ (5)	95	56
$\text{Rh}_2(\text{S-DOSP})_4$ (5) ^d	89	86

^aReactions were performed as described in Table 1. ^bYield after of removal of catalyst. ^cAnalysis on a WHELK-O column using 20% EtOAc in hexanes. ^dReaction performed in pentane.



Scheme 4

Acknowledgement

Support for this research from the National Science Foundation and from the National Institutes of Health (GM-46503) is gratefully

acknowledged. We are grateful to Professor Huw M. L. Davies for providing us with a generous sample of $\text{Rh}_2(\text{S-DOSP})_4$. This letter is dedicated to Professor Ryogi Noyori, whose insightful contributions to the development of asymmetric catalysis for metal carbene transformations has led us to fruitful discoveries.

References and Notes

- (1) Doyle, M. P.; McKervey, M. A.; Ye, T. *Modern Catalytic Methods for Organic Synthesis with Diazo Compounds: From Cyclopropanes to Ylides*; Wiley, New York, 1998.
- (2) Doyle, M. P.; McKervey, M. A. *J. C. S. Chem. Commun.* **1997**, 983. Doyle, M. P.; Forbes, D. C. *Chem. Rev.* **1998**, *98*, 911.
- (3) Padwa, A.; Austin, D. J. *Angew. Chem. Int. Ed. Engl.* **1994**, *33*, 1797. Padwa, A.; Krumpke, K. E. *Tetrahedron* **1992**, *48*, 5385.
- (4) Davies, H. M. L. *Eur. J. Org. Chem.* **1999**, 2459.
- (5) Doyle, M. P.; Westrum, L. J.; Wolthuis, N. E.; See, M. M.; Boone, W. P.; Bagheri, V.; Pearson, M. M. *J. Am. Chem. Soc.* **1993**, *115*, 958.
- (6) Taber, D. F.; Ruckle, R. E., Jr. *J. Am. Chem. Soc.* **1986**, *108*, 7686.
- (7) Doyle, M. P.; Kalinin, A. V.; Ene, D. G. *J. Am. Chem. Soc.* **1996**, *118*, 8837.
- (8) Anada, M.; Hashimoto, S. *Tetrahedron Lett.* **1998**, *39*, 79.
- (9) Wee, A. G. H. *Tetrahedron Lett.* **2000**, *41*, 9025.
- (10) Wang, P.; Adams, J. *J. Am. Chem. Soc.* **1994**, *116*, 3296.
- (11) Davies, H. M. L.; Hansen, T.; Churchill, M. R. *J. Am. Chem. Soc.* **2000**, *122*, 3063.
- (12) Imai, T.; Nichida, S. *J. Org. Chem.* **1979**, *44*, 3574.
- (13) Doyle, M. P.; Zhou, Q.-L.; Simonsen, S. H.; Lynch, V. *Synlett* **1996**, 697.
- (14) Doyle, M. P.; Davies, S. B.; Hu, W. *Org. Lett.* **2000**, *2*, 1145.
- (15) Davies, H. M. L.; Hansen, T.; Hopper, D.; Penaro, S. A. *J. Am. Chem. Soc.* **1999**, *121*, 6509.
- (16) Davies, H. M. L.; Stafford, D. G.; Hansen, T. *Org. Lett.* **1999**, *1*, 233.
- (17) Hoppe, I.; Schöllkopf, U. *Liebigs Ann. Chem.* **1979**, 219.
- (18) Black, T. H.; DuBay, W. J.; Tully, P. S. *J. Org. Chem.* **1988**, *53*, 5922.
- (19) Diastereoisomers were separated, and they were identified by spectral methods; assignments were based on nOe experiments.

Article Identifier:

1437-2096,E;2001,0,SI,0967,0969,ftx,en;Y02801ST.pdf